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New Lactimidomycin Congeners Shed Insight into Lactimidomycin Biosynthesis in *Streptomyces* amphibiosporus

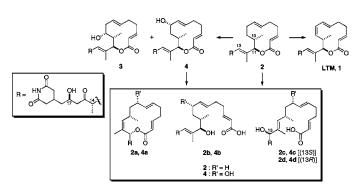
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ABSTRACT



Lactimidomycin (LTM, 1) is a macrolide antitumor antibiotic with a glutarimide side chain from *Streptomyces amphibiosporus* ATCC53964. To further develop LTM and related analogues as drug candidates we have (i) improved LTM production by \sim 20 fold, (ii) identified three new metabolites (2–4) possibly involved in the LTM biosynthetic pathway; (iii) found 3 to be identical with a previously identified isomigrastatin precursor, (iv) determined the absolute stereochemistry of LTM, and (v) produced new LTM rearrangement products 2a–d and 4a–d.

Lactimidomycin (LTM, 1), a macrolide antibiotic with a glutarimide side chain containing an unsaturated 12-membered lactone ring, was first discovered in 1992 from the fermentation broth of *Streptomyces amphibiosporus* ATCC-53964, with its stereochemistry undetermined. LTM exhibited strong in vitro cytotoxicity against a number of human cell lines (IC $_{50} = 3.0-65$ nM) and in vivo antitumor activity in mice. In addition, it showed antifungal activity and inhibited both DNA and protein syntheses. LTM belongs to a small group of glutarimide-containing polyketide natural products; others include streptimidone, cycloheximide, cycloheximide,

migrastatin (MGS, **5**),⁴ dorrigocin (DGN) A (**6a**) and B (**7**),⁵ NK30424A/B (**8**),⁶ and iso-migrastatin (iso-MGS, **9**)⁷ (Figure 1). We previously found that **5**–**7** are shunt metabolites of

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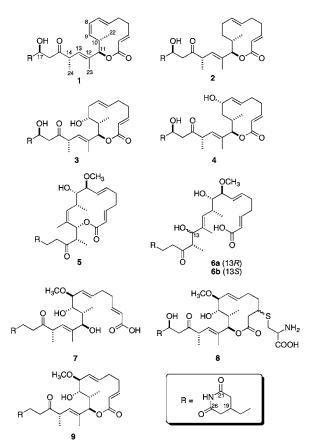


Figure 1. Representative glutarimide-containing polyketide natural products.

9, and that congeners of 9 undergo H₂O-mediated rearrangements to produce the linear dorrigocin and ring-expanded migrastatin scaffolds.^{8,9} Iso-MGS and its congeners also undergo facile thermal rearrangements to afford the migrastatin scaffold exclusively.¹⁰ We also optimized the fermentation conditions of *Streptomyces platensis*, the producer of 9, which led to the isolation and characterization of numerous biosynthetic intermediates, shedding light on the post-polyketide synthase (PKS) tailoring steps for biosynthesis of 9.9

Since 1 and 9 share several structural features, we further examined the fermentation behavior of *S. amphibiosporus* to elucidate LTM biosynthesis and to possibly identify new analogues of 1. Such analogues might provide clues to the biosynthetic relationship between 9 and 1 in addition to possessing useful bioactivities of their own.

Here we report the following: (i) fermentation optimization of *S. amphibiosporus* leading to dramatically improved yields of LTM (isolated yield: 80 mg/L), (ii) isolation and structure determination of three structurally related homo-

logues of LTM (Figure 1), namely, 8,9-dihydro-LTM (2), 8,9-dihydro-9-hydroxy-LTM (3), and 8,9-dihydro-8-hydroxy-LTM (4) that provide valuable clues to the post-PKS tailoring steps in LTM biosynthesis, (iii) the structure of 3 is identical with that of a previously reported precursor to 9, (iv) establishment of the absolute stereochemistry for 1 and its three congeners, and (v) H_2O -mediated rearrangements of 2 and 4 resulting in semi-synthesis of linear dorrigocin and ring-expanded migrastatin scaffolds; related rearrangement products for 3 have been previously reported.⁹

It is well-known that alteration of the cultivation parameters (for example, media composition, aeration, culture vessel, pH value, etc.) can affect the metabolic profile of various microorganisms.¹¹ We started with fermentation of *S. amphibiosporus* using a modified medium composition (Supporting Information). After fermentation, the culture broth was extracted with EtOAc. HPLC-UV and LC-MS analyses of the crude extract revealed the presence of **1** as the only metabolite related to biosynthesis of **1** (Figure 2A).

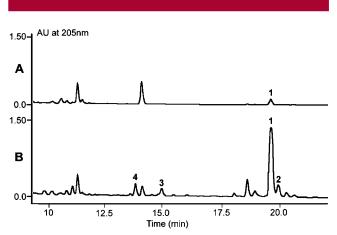


Figure 2. HPLC profiles of (A) EtOAc fermentation extract in the absence of resin and (B) EtOH eluent of XAD-16 adsorbent resin harvested from *S. amphibiosporus* fermentation broth. Peak assignments correlate to compounds shown in Figure 1. Other major peaks are metabolites unrelated to **1** biosynthesis.

Scale-up fermentation (3 L) gave an isolated yield of approximately 3.0 mg/L, which is comparable to the isolated yield reported previously. We then introduced adsorbent resin to the production medium because of its ability to sequester, stabilize, and control the distribution of the secondary metabolites. Not surprisingly, HPLC-UV and LC-MS analyses of the resin extract suggested the titer of LTM was dramatically improved when 2.0-3.5% (w/v) XAD-16 adsorbent resin was supplemented in the production medium. Moreover, three additional LTM analogues were observed (Figure 2B). High percentages of resin (>4%) in the culture medium of *S. amphibiosporus* led to reduced or no production of LTM, suggesting that the strain is intolerant of high resin percentages. Large-scale fermentation of *S. amphibiosporus* (20 L) in the presence of 3% XAD-16 resin

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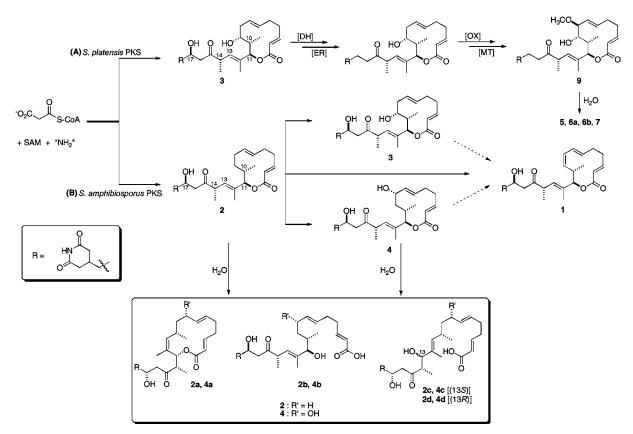


Figure 3. Compounds 2 and 3 en route to 1, 5–7 biosynthesis. (A) 3 as the nascent polyketide product of the *S. platensis* PKS and precursor to 9 via dehydration, enoyl reduction, oxidation, and methylation. Iso-MGS (9) is a precursor to 5 and DGNs 6a, 6b, and 7. The timing of steps involved in 9 production from 3 is not yet determined. (B) 2 as the nascent polyketide product of the *S. amphibiosporus* PKS and precursor to 1, 3, and 4. Both 3 and 4 could be potential precursors to 1. Intermediate 3 is also a precursor to 9 as noted in pathway A, and as such, 3 is an important branch point that co-relates compounds 1, 5–7, and 9. Compounds 2 and 4 rapidly undergo H₂O-mediated rearrangements to provide 2a–d and 4a–d as has been observed for 3, 9, and other analogues. Sepanda Sepa

was conducted. After fermentation, the resin was collected by filtration with cheesecloth, air-dried, and then eluted with absolute ethanol. Solvent removal followed by silica gel chromatography with CHCl₃/MeOH and EtOAc/hexane afforded compounds **1–4** in isolated yields of approximately 80, 3, 5, and 5 mg/L, respectively.

Compound **3** was obtained as a colorless oil with MS, ¹H, and ¹³C NMR spectra (measured in CDCl₃) and optical rotation identical with those of 17-hydroxy-8-desmethoxy-iso-MGS isolated recently by us from *S. platensis*, whose absolute stereochemistry (9*S*,10*S*,11*R*,14*S*,17*R*) has been established (Supporting Information). We have previously shown that **3** can undergo H₂O-mediated rearrangement to the corresponding MGS and DGN analogues. Isolation of **3** from both *S. platensis* and *S. amphibiosporus* suggests that **9** and **1** share similar biosynthetic machineries for construction of C-10, C-11, C-14, and C-17 (Figure 3).

The NMR spectral data of 1 were originally reported in DMSO- d_6 , leaving its relative and absolute stereochemistry unsolved. We recently established the absolute stereochemistry of 9 and its congeners. It therefore seemed reasonable that the absolute stereochemistry of 1 at C-10, C-11, C-14, and C-17 might be the same as that observed for analogues

of **9**. To verify this hypothesis, we acquired 1D and 2D NMR spectral data of **1** in CDCl₃, which enabled complete accounting for all ¹H and ¹³C NMR signals (Supporting Information). The relevant ¹H and ¹³C NMR chemical shifts and proton splitting centers at C-10, C-11, C-14, and C-17 are as found in **3** and 17-hydroxy-iso-MGS. ⁹ Thus, the absolute stereochemistry of **1** was assigned as 10*S*,11*R*,14*S*,17*R*.

Compound 2 was purified as a colorless oil having a molecular formula of C₂₆H₃₇O₆N. HR-MALDI-MS reveals that 2 has a mass 2 amu greater than 1, corresponding to a saturated double bond equivalent. The ¹³C NMR spectrum of 2 displayed 26 carbon signals, 6 of which were salient olefinic carbons, suggesting 2 as the dihydro-derivative of 1. Analyses of its 1D and 2D NMR (¹H-¹H COSY, TOCSY, HMQC, and gHMBC) spectra allowed full assignments of its ¹H and ¹³C signals (see the Supporting Information), verifying saturation of the C8/C9 in 2. The COSY correlations between H-10/H₂-9/H₂-8/H-7, and gHMBC correlations from H-11/C-9, H-22/C-9, H-10/C-9/C-8, and H-7/C-9/ C-8, confirmed the above conclusion. Other 1D and 2D NMR spectral data of 2 agreed with the rest of the structure and confirm that the stereochemistry of 2 is identical to that of 1 and 3.

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Compound 4 was obtained as a white powder and was found to have the same molecular formula, C₂₆H₃₇O₇N, as 3 provided by high-resolution MALDI-MS. The ¹H and ¹³C NMR (in CDCl₃) spectra of 4 showed significant similarities to those of 3. To clarify this situation, a series of 2D NMR (1H-1H COSY, TOCSY, HMQC, and gHMBC) spectra of 4 were acquired, which allowed full assignment of all ¹H and ¹³C signals (Supporting Information). Analysis of ¹H and ¹³C NMR data for 4 revealed the presence of the same C-11 side chain found in 1-3. The COSY correlations between H_2 -16/H-17 (δ 4.05)/ H_2 -18 confirmed the hydroxy group at C-17. In the gHMBC spectrum of 4, the signal due to H-11 $(\delta 5.19)$ showed correlations with C-1 $(\delta 167.4)$, C-12 $(\delta$ 134.9), C-13 (\delta 128.5), C-9 (\delta 39.4), C-10 (\delta 32.5), C-22 $(\delta 19.8)$, and C-23 $(\delta 15.2)$, establishing the 12-membered lactone structure. The COSY correlations between H-7 (δ 5.27)/H-8 (δ 4.45)/H₂-9 (δ 1.51, 1.97) and gHMBC correlations between H-7/C-8 (δ 73.4) and H-9 (δ 1.51)/C-8 placed another hydroxyl group at C-8. Additional examination of the 1D and 2D NMR data permitted complete structural elucidation of 4 to be made with the exception of stereochemistry (Figure 1).

The absolute stereochemistry of 4 was assigned by comparing its ¹H and ¹³C NMR chemical shifts and proton splitting patterns with those of 3 and iso-MGS congeners. The configuration of the two double bonds in the lactone ring was assigned as trans due to the large coupling constant of 15.5-16.0 Hz. As in the case of 1, 4 has the same stereochemistry at C-10, C-11, C-14, and C-17 on the basis of the relevant proton chemical shifts, coupling constants (e.g., $J_{\text{H}10,\text{H}11} = 5.0 \text{ Hz}$, $J_{\text{H}13,\text{H}14} = 9.5 \text{ Hz}$), and the close ¹³C NMR chemical shifts from C-10 to C-19. The coupling constant between H-7 and H-8 (J = 8.0 Hz) is significantly different from that of 9 (J = 3.5 Hz), which has an 8β -OCH₃ configuration, indicating an 8α-OH configuration in **4**. This scenario is also consistent with the $J_{\rm H7,H8\beta}$ value (7.0-9.5 Hz) observed in 8-desmethoxy-iso-MGS and its analogues.9 The 8-OH configuration in 4 was further supported by the observation of the dramatic ¹H and ¹³C NMR signal downfield shifts of 22-CH₃ (Δ0.14 ppm and $\Delta 11.4$ ppm, respectively) when compared with those of 3, due to a syn-parallel disposition. Hence, the absolute stereochemistry of 4 was assigned as 8S,10S,11R,14S,17R.

Iso-MGS congeners including **3** are relatively stable in most organic solvents. However, in water, they undergo H₂O-mediated rearrangement into their corresponding 14-membered migrastatin and ring-opened dorrigocin analogues (Figure 3). Gompounds **2** and **4** also possess similar chemical properties; incubation of **2** and **4** in H₂O-DMSO (9:1) at 37 °C for 12 h irreversibly resulted in a series of new substances (Figure 3). Rearrangement products **2a-d** and **4a-d** (Figure 3) correspond to migrastatin/dorrigocin counterparts and are the overwhelmingly predominant products observed by HPLC, LC-MS, and NMR analyses (Supporting Information). Similar, but reversible, rearrangements have been observed in a recent example describing the ring-opened form of pseudotrienic acids A and B from *pseudomonas sp.* MF381-IODS. These acids were found to

readily interconvert with their respective macrolide forms.¹² Under identical conditions, **1** was found to be stable and remained unchanged. These results provide experimental evidence supporting the idea that the C8/C9 double bond changes the 12-membered-ring configuration thereby influencing the propensity to undergo ring-opening and ring-expansion rearrangements.

Feeding studies with ¹³C-labeled sodium acetate and methionine have shown that LTM is of polyketide origin.¹ The isolation of 2-4 implies that 2 is the nascent polyketide product assembled by the PKS and that 3 and 4 may be shunt metabolites formed during the desaturation of 2 en route to 1 (Figure 3). Desaturation has been elaborately involved in natural products biosynthesis and drug metabolism. For instance, desaturation and hydroxylation of valproic acid has been demonstrated by using a purified and reconstituted microsomal cytochrome P-450 in vitro. 13,14 Similarly, testosterone is known to undergo P-450-catalyzed oxidation.¹⁵ In both cases, substrate oxidation provides one olefinic product and two hydroxylated products. We believe that LTM provides another example of this unique pathway in which a fully saturated C-C bond (in 2) is converted to two hydroxylated congeners 3 and 4 and the diene-containing LTM (1). This notion is promoted by our ability to detect only one stereoisomer of each hydroxylated regioisomer. The stereospecificity of hydroxylation suggests that this process is enzyme-catalyzed. This is significant since the chemistry carried out appears to occur following the completion of all PKS orchestrated steps. Such post-PKS tailoring events have not yet been noted in the biosynthesis of glutarimidecontaining polyketide natural products. Thus, our disclosure of compounds 2-4 here sheds new insight into LTM biosynthesis.

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Supporting Information Available: Experimental procedures, MS and ¹H and ¹³C NMR data of **1**, **2**, **4**, **2a**–**c**, and **4a**–**c**, copies of ¹H and ¹³C NMR spectra of **2** and **4**, optical rotation data for **2**–**4**, and MS data for **3**. This material is available free of charge via the Internet at http://pubs.acs.org.

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